

## Biosorption of Chromium by Living Cells of *Azotobacter s8*, *Bacillus subtilis* and *Pseudomonas aeruginosa* using Batch System Reactor

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### ABSTRACT

Chromium in wastewater is classified as one of the dangerous contaminants that require further treatment before being discharged to water body. The concentration of chromium in water body, especially river, has increased as many industries utilize chromium as raw material and then discharge their wastewater without any treatment. Biosorption is one of methods that are widely used to treat heavy metal containing wastewater. Bacteria are the most common microorganisms to be used as heavy metal treatment agent. *Azotobacter s8*, *Bacillus subtilis* and *Pseudomonas putida* had been proven to have a heavy metal resistant capability. The screening test showed that Minimum Inhibitory Concentration (MIC) value of chromium for all bacteria ranged from 100 to 250 mg/L of CrCl<sub>3</sub>. The chromium biosorption test by bacteria showed that *Azotobacter s8* was able to remove 10.53%, and *Bacillus subtilis* was able to remove 5.68% chromium from 50 mg/L initial concentration, while *Pseudomonas putida* showed no chromium removal. The chromium biosorption capacity by *Azotobacter s8* was 580.08 mg/g and 349.30 mg/g for *Bacillus subtilis*.

**Keywords:** *Azotobacter*, *B. subtilis*, Bacteria, Biosorption, Cr, *P. Putida*

### INTRODUCTION

Chromium is one of heavy metals commonly found in water bodies as the result of industrial activity. Usually, chromium comes from the electroplating industry, metal industry, and tanning industrial effluent [Oves et al., 2013; Zahoor and Rehman, 2009]. Chromium exists in nature in several oxidation numbers, from -2 to +6 [Evelyne and Ravisankar, 2014]. However, there are only 2 commonly discovered forms, which are Cr<sup>3+</sup> and Cr<sup>6+</sup> [Kaur et al., 2014]. The Cr<sup>6+</sup> is the more unstable form because it can react directly with other particles in the air to form Cr<sup>3+</sup> [Evelyne and Ravisankar, 2014]; therefore most treatment alternatives are designed for its more stable form, Cr<sup>3+</sup> [Sundar et al., 2011].

The chromium concentration in water bodies usually ranges from 0.1 to 6 mg/L [Duman et al., 2009]. Those concentrations can increase when there are pollutants entering a water body. The chromium concentration in wastewater ranges from 10 to 25,000 mg/L [Mythili and Karthikeyan, 2011]. In turn, the chromium concentration permitted for drinking water is  $\leq 0.05$  mg/L, set by Ministry of Health Indonesia (2010) and for wastewater it is  $\leq 0.5$  mg/L set by Minister of Environment Indonesia (2014). The existing concentration is far exceeding the standard for both drinking water and wastewater. Therefore, further treatment is needed to handle this condition [Mythili and Karthikeyan, 2011].

Biosorption is a method of treating the chromium-containing wastewater [Evelyne and Ravisankar, 2014]. The principal of biosorption

process is to utilize the microorganism ability in protecting themselves from heavy metal exposure [Kaur et al., 2014]. Previous studies showed that *Azotobacter s8* [Purwanti et al., 2016], *Bacillus sp.* [Mythili and Karthikeyan, 2011] and *Pseudomonas sp.* [Kaur et al., 2014] can remove  $\text{Cr}^{3+}$  from wastewater. The three bacteria can be isolated from the chromium-containing wastewater. Chromium removal was usually studied with the contact duration ranging from 1 to 7 days and conducted in mixed culture bacteria [Deepali, 2011]. To the best of our knowledge, the study about the utilization of single culture from three types of those bacteria in short contact duration, which is the exponential phase of the bacterial growth, has not been conducted yet. This study aimed at analyzing the chromium biosorption potential in batch reactor using living cells of 3 single culture of bacteria, which were *Azotobacter s8*, *Bacillus subtilis* and *Pseudomonas putida* in short contact duration, which was during bacterial exponential growth. These presented results may give a new view in the field of chromium removal by single culture of bacteria in a very short contact time.

## MATERIALS AND METHOD

### Chromium stock solution preparation

The chromium stock solution was made by using Chromium (III) Chloride (Merck, Germany). A certain amount of chromium powder was diluted in a previously autoclaved aquadest (OneMed, Indonesia) to achieve the desired concentration [Purwanti et al., 2017]. The solution would be used in the main study as the substrate to be removed.

### Bacterial regrowth

Bacterial regrowth was conducted before performing the growth test and chromium removal test. The method used in bacterial regrowth was adapted from Machmud (2001) with some adjustments. At this stage, the tested bacteria were regrown to keep them from being contaminated. Regrowth also provided additional bacteria stock in case of failure or additional needs. The three tested bacteria, *Azotobacter s8*, *Bacillus subtilis* and *Pseudomonas putida* were moved from the laboratory culture medium to a new slant Nutrient Agar (NA) medium (Merck, Germany). After the moving, the bacteria were incubated in incubator (Memert+, Germany) for 24 hours.

### Bacterial growth rate test

Bacterial growth rate test provided the information regarding the bacterial growth duration and determined the exponential phase [Kurniawan et al., 2018]. The exponential phase time would then be used as the determinant time for chromium (Cr) removal test. The method in this test was adapted from Deepali (2011) with some adjustments. The bacterial growth rate test was conducted for 24 hours in 250 ml Erlenmeyer filled with Nutrient Broth (Merck, Germany) medium on a rotary shaker (V-tech, USA). The 600nm Optical Density ( $\text{OD}_{600}$ ) was checked every 2 hours for 24 hours. The bacterial growth curve was made by plotting measured  $\text{OD}_{600}$  vs time.

### Minimum Inhibitory Concentration (MIC) test

Minimum inhibitory concentration test was carried out to determine the minimum concentration of chromium that could totally inhibit the bacterial growth. The method used in MIC test was adapted from Mythili and Karthikeyan (2011) and Ruangpan (2004), with some adjustments. In MIC test, three tested bacteria were cultured on the chromium-contaminated NA medium with the concentration of 0, 5, 50, 100, 250, and 500 mg/L. The screening method was used, which involved streaking onto the surface of chromium-contaminated NA medium. The control that was used at this stage was NA medium without any chromium content.

The growth of bacteria on the surface of chromium contaminated NA was observed after 24 hours of 37°C incubation. The growth of bacteria on the medium with chromium content was then compared to the growth on control. The MIC value was determined by visual observation. The MIC value was concluded by the lowest concentration at which there is no bacterial growth at all. After the MIC value was determined, one concentration below the MIC value would be chosen as tested concentration in chromium removal test.

MIC scoring was determined based on the area of bacterial growth on the chromium-contaminated medium, compared to the bacterial growth on control. Scoring was displayed in + and – symbol with the complete scoring as follows:

+++++ = Bacterial growth area compared to control was 81–100%

++++ = Bacterial growth area compared to control was 61–80%

- +++ = Bacterial growth area compared to control was 41–60%
- ++ = Bacterial growth area compared to control was 21–40%
- + = Bacterial growth area compared to control was ≤ 20%.
- = No bacterial growth

### Chromium removal test

The removal test was performed to determine the chromium removal percentage by bacteria towards the given chromium concentration. The method used in this stage was adapted from Deepali (2011), with some adjustments. This test also determined the influence of bacterial species towards the removal of chromium. At this stage, the analyzed parameters were the number of bacterial colonies and total chromium concentration. The chromium removal test was conducted in 250 mL Erlenmeyer reactor filled with 100 mL chromium solution as living medium in batch system. A 2% of living bacteria culture ( $OD_{600} = 1A$ ) was inoculated into reactor in the beginning of the test period.

All parameters were analyzed 2 times, at the start and the end of the experiment. The  $Cr^{3+}$  concentration used was the one determined from MIC test [Ruangpan, 2004]. The number of bacterial colonies was analyzed using standard plate method to determine the bacterial growth and its weight during chromium removal test. The test period at this stage was bacterial exponential growth time obtained from the bacterial growth rate test. The bacteria in this study consisted of 3 species i.e. *Azotobacter s8*, *Bacillus subtilis* and *Pseudomonas putida*. The chromium biosorption capacity by bacteria was determined by using

indirect mathematical calculation. All analyses at this stage were conducted in duplicate.

## RESULTS AND DISCUSSION

### Bacterial growth rate

Figure 1 indicates that the bacterial acclimatization phase occurred from hour 0 until hour 2, and continued with exponential growth phase from hour 2 to hour 6. In accordance with Azoddein et al., 2017; Purwanti et al., 2016; and Ramasamy et al., 2015, the exponential growth rate of *Azotobacter*, *Bacillus sp.*, and *Pseudomonas sp.* occurred on hour 6 of observation. Exponential phase of all bacteria was followed by stationary phase from hour 6 to the end of test time. It can be seen that no death phase is obtained in the bacterial growth curve, which happened due to the indirect method of predicting the growth of bacteria used in this research. Dwipayana and Ariesyady (2012) stated that the OD measurement was an indirect method of determining the bacterial growth. This method cannot distinguish between the living and dead bacteria; thus, the OD always tend to be higher until the end of test period. From the bacterial growth rate test, the test duration that would be used for chromium biosorption stage was 6 hours.

### Minimum Inhibitory Concentration (MIC)

Table 1 shows that bacterial growth declined along with the increase of the  $Cr^{3+}$  concentration on the medium. At the concentration of 5 to 50 mg/L of  $CrCl_3$ , all three tested bacteria showed good growth compared to the control. All tested

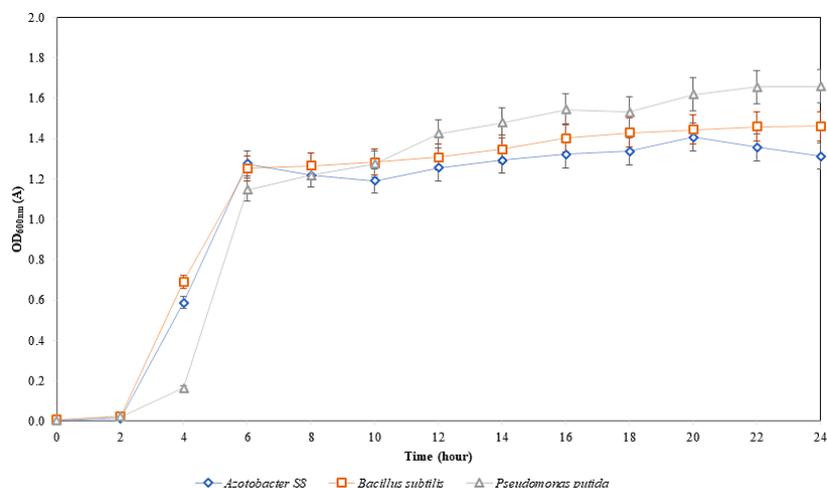


Figure 1. Bacterial growth rate

bacteria showed significantly declined growth at the of concentration 100 mg/L CrCl<sub>3</sub> (P < 0.05). This decline was obvious when compared with the growth of the control medium. The growth declining showed the inhibition caused by Cr<sup>3+</sup> in the medium [Kurniawan et al., 2018]. The tested bacteria did not show any growth at the concentration of 250 mg/L and 500 mg/L of CrCl<sub>3</sub>. No growth indicated a full inhibition by Cr<sup>3+</sup> to the bacterial metabolism [Ruangpan, 2004].

All bacteria exhibited very good growth at the concentration of 5 mg/L CrCl<sub>3</sub>. This showed that all tested bacteria had considerably high tolerance on that concentration. All bacteria showed good growth at the concentration of 50 mg/L and poor growth at the concentration of 100 mg/L CrCl<sub>3</sub>. This showed an evidence that bacterial growth started to stunt along with the increasing of CrCl<sub>3</sub> [Titah et al., 2018], up to concentration of 100 mg/L. All tested bacteria did not show any growth starting from the concentration of 250 mg/L CrCl<sub>3</sub>. It could be concluded that the MIC of all three bacteria ranged from 100 to 250 mg/L of CrCl<sub>3</sub>. For chromium removal test, one Cr<sup>3+</sup> concentration was chosen under MIC range value [Ruangpan, 2004], which was 50 mg/L CrCl<sub>3</sub>. This concentration was chosen in consideration that the bacterial growth at this concentration was still good compared to the control.

### Chromium removal and biosorption

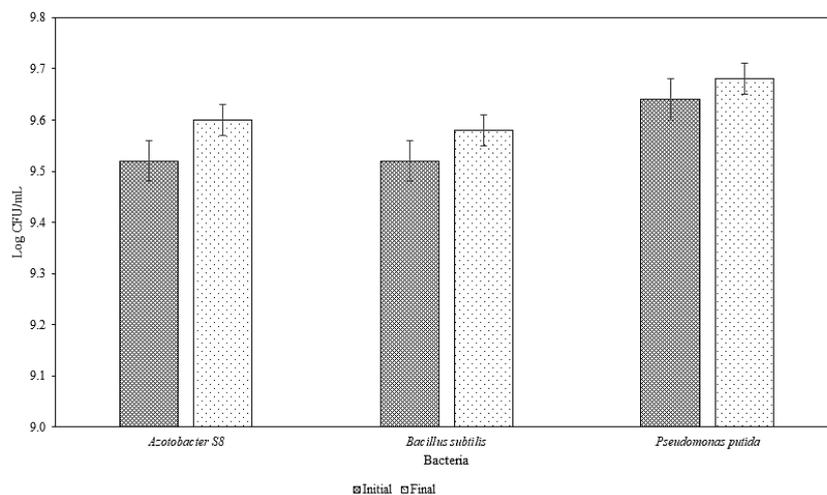
The results of the analysis on the number of bacterial colonies (Figure 2) showed that all bacteria in the test reactor experienced growth. The measured bacterial growth in all reactors was not significant (p > 0.05). This was normal considering that the test period used was during exponential growth phase, which was only 6 hours [Ramasamy et al., 2015]. Colony increase during chromium removal test was then converted into weight unit to be used as bio-sorbent weight of chromium removal. The converted weight from bacterial colonies can be seen in Table 2 with colony weight conversion calculated using the following formula [Deriase and El-Gendy, 2014]:

$$DCW(g/L) = 0.55147 \left( \frac{CFU/mL}{4.25 \times 10^{12}} \right)^{0,0539} + 0.33042 \quad (1)$$

The chromium removal result (Figure 3) showed a decrease of chromium concentration in *Azotobacter s8* and *Bacillus subtilis* reactor. In the reactor of *Pseudomonas putida*, there was no chromium removal. This indicated that in the exponential growth of 6 hours, *P. putida* was not able to perform its metabolism normally [Azoddein et al., 2017]. *Azotobacter s8* showed

**Table 1** Scoring of MIC

Bacterial Species	Concentration of Chromium (mg/L)					
	0	5	50	100	250	500
<i>Azotobacter s8</i>	+++++	+++++	++++	++	-	-
<i>Bacillus subtilis</i>	+++++	+++++	++++	+++	-	-
<i>Pseudomonas putida</i>	+++++	++++	+++	++	-	-



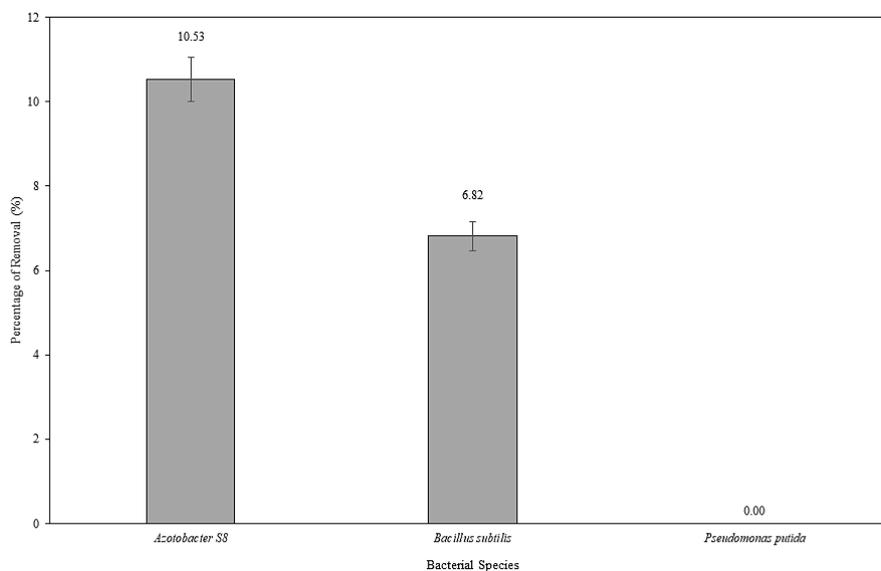
**Figure 2** Number of bacterial colonies during chromium removal test

**Table 2.** Conversion result of bacterial colonies weight

Bacterial Species	Dry Cell Weight during Chromium Removal Test (g/L)
<i>Azotobacter s8</i>	0.431
<i>Bacillus subtilis</i>	0.429
<i>Pseudomonas putida</i>	0.427

**Table 3.** Biosorption capacity for each bacterial species

Bacterial Species	Biosorption Capacity (mg/g)
<i>Azotobacter s8</i>	580.08
<i>Bacillus subtilis</i>	349.30
<i>Pseudomonas putida</i>	0.00

**Figure 3.** Chromium removal

10.53% removal and *B. subtilis* 6.82% removal of chromium.

The amounts of chromium removed by the two bacteria during removal test were 2.5 mg/L for *A. s8* and 1.5 mg/L for *B. subtilis*. The removed chromium during the removal test was then used as substrate weight in the calculation of biosorption capacity [Deriase and El-Gendy, 2014]. The biosorption capacity for each bacterium was calculated using the following equation and the result was shown in Table 3.

Table 3 shows that *A. s8* had the adsorption capacity of 580.08 mg/mg, *B. subtilis* was 349.30 mg/g and *P. putida* was 0 mg/mg. In accordance with García et al. (2016) and Purwanti et al. (2016), *A. s8* and *B. subtilis* have a capability to perform biosorption of heavy metal. The calculation showed that *A. s8* had the highest biosorption capacity among all the tested bacteria.

From this result, the scale up for *A. s8* and *B. subtilis* for greater pollutant volume and concentration could be estimated. Since this treatment used bacteria as chromium removal agent, the biomass requirement as bio-sorbent need to

be concerned and calculated carefully before applying it in the industrial sector [Gomes et al., 2018]. The treatment of biomass after absorbing chromium can be an interesting topic for further research. Moreover, the result of this result was still limited to the scale up in the terms of the pollutant volume and concentration, an extent in terms of time cannot be predicted using the results of this research.

## CONCLUSION

At 6 hours of exponential growth time, *Azotobacter s8* showed the highest percentage in chromium removal, up to 10.53% and 580.08 mg/mg in term of biosorption capacity. *Bacillus subtilis* showed 6.82% removal and 349.30 mg/mg biosorption capacity of chromium. *Pseudomonas putida* showed no removal of chromium; therefore, the biosorption capability could not be calculated. *Azotobacter s8* had a good potential to be used as biosorption agent in further research of chromium removal in wastewater.

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